Alpha-oxoketenes from HPALDs efficiently convert HO₂ into OH radicals in isoprene oxidation

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4-Hvdroperoxy-2/3-methyl-but-2-enals (HPALDs) have been predicted to be produced by 1.6-H shift isomerisation of Z-δ-hydroxy-isoprenylperoxy radicals in the atmospheric oxidation of isoprene by OH at low NO.^{1,2} They have recently been observed and recognized as important intermediates in isoprene oxidation,³ although the peroxy isomerisation rate and HPALD yield as well as the contribution to OH recycling still remain very uncertain.^{3,4,5} The HPALD formation and subsequent photochemistry involving very fast photolysis processes have been proposed to result overall in 2 HO₂ and up to 3 OH radicals, with α -oxoketenes as main products.² On the other hand, recent modeling studies suggest that the very high OH observed in isoprene-rich areas could be explained (partly) by unknown compounds reacting fast with HO₂ and converting it into OH.^{6,7} However, at present no VOCs or OVOCs are known to react fast with HO₂. In this work, the reactions of HO₂ with the α -oxo-ketenes from the HPALDs have been investigated theoretically. The potential energy surfaces, established using suitable DFT/ab initio methods, reveal that HO₂ adds to the α -oxo-ketenes in a three-step process over submerged transition states, forming a peroxy radical. The theoretically estimated rate constants are sufficiently high that reaction with HO₂ should be the predominant removal route of the oxoketenes in the conditions of interest. The straightforward subsequent chemistry of the peroxy product is expected to result partly in the removal of a second HO₂, and to yield overall 1 to 2 OH radicals, with methylglyoxal, CO₂, CO, and acetylperoxy radical as main co-products. Modeling results, using the IMAGES CTM, are presented on the impact of these reactions on OH recycling and HO₂ levels in isoprene-rich areas for various scenarios regarding the isopreneperoxy isomerisation rates and the yields of the HPALDs and their α -oxoketene products.

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