Fundamentals of Chemical ReactivityInvitedWednesday 9:00 - 9:40

Modeling Chemical Reactions in the 21st Century: Picking the Right Method AND the Right Problem*

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Theoretical tools for modeling chemical reactions have become quite highly developed and diverse. This confronts us with many choices in tailoring the theoretical description of a reaction to the requirements of specific applications. We shall review several reactions we have recently treated that required different levels of theory: $F+HCl\rightarrow HF+Cl(1)$, $HO_2+HO_2\rightarrow H_2O_2+O_2(2)$, $CH_2FOH\rightarrow CH_2O+HF(3)$, and $CH_3OH+HO_2\rightarrow CH_3O+H_2O_2(4)$. We shall emphasize the role of reactive complexes, torsional motion of the collision complex, and the possibility of water photo-catalysis in the discussion. In addition to selecting the appropriate method to model a given reaction, we also need to face the question of determining which reaction in a given mechanism is the most important to investigate. We shall describe some recent advances in global sensitivity analysis for identifying key reactive steps in large mechanisms.

* Collaborative work, in part, with MJ Davis, V Vaida, L Harding, and K Han

References

- (1) R. T. Skodje, "Resonances in Bimolecular Chemical Reactions", Adv. Quantum Chem. 63, 119 (2012).
- (2) D. D. Y. Zhou, K. L. Han, P. Y. Zhang, L. B. Harding, M. J. Davis, and R. T. Skodje, "Theoretical Determination of the Rate Coefficient for the HO₂+HO₂→H₂O₂+O₂ Reaction: Adiabatic Treatment of Anharmonic Torsional Effects", J. Phys. Chem. A, submitted.
- (3) Z. C. Kramer, K. Takahashi, and R. T. Skodje, "Water Catalysis and Anticatalysis in Photochemical Reactions: Observation of a Delayed Threshold Effect in the Reaction Quantum Yield", J. Am. Chem. Soc. 132, 15154 (2010).
- (4) R. T. Skodje, A. S. Tomlin, S. J. Klippenstein, L. B Harding, and M. J. Davis, "Theoretical Validation of Chemical Kinetic Mechanisms: Combustion of Methanol", J. Phys. Chem. A. 114, 8286 (2010).



0.8 Stage 1 CH3OH+HO2 (Global) 0.4 HO2+HO2 сн₃он+о₂ 02+N ŝ 0.0 40 80 Index i (Local) -0.4 ŝ -0.8

Resonance structure of the F+HCL reaction

Global sensitivity analysis of methanol combustion