

Reactive Nitrogen, Ozone, and Radicals

Glenn M. Wolfe (Moderator)
NASA Goddard Space Flight Center

AGES Meeting
28 September 2022

Speaker	Measurement	Mission
Glenn (for Lee Mauldin)	RO _x , H ₂ SO ₄	GOTHAAM
Glenn (for Ale Franchin)	NO _x , O ₃	GOTHAAM
Andrew Rollins	NO _x , NO _y , O ₃	AEROMMA
Ilana Pollack	NH ₃	AEROMMA
Aaron Stainsby	OH Reactivity	AEROMMA
Saewung Kim	OH Reactivity	NYC-METS
Ezra Wood	HO ₂ + RO ₂	NYC-METS
Trevor VandenBoer	Acids, Total N _r	THE CIX
Paul Walter	Sondes	STAQS
Steve Brown	NO _y , O ₃	CUPiDS
Steve Brown	“Night NO _x ”	AEROMMA

OH, HO₂, HO₂+RO₂, H₂SO₄ Measurements
Lee Mauldin, Chris Cantrell, Emmanuel Assaf
University of Colorado, Boulder



OH

OH is measured by first converting it to $^{34}\text{H}_2\text{SO}_4$ via reaction with $^{34}\text{SO}_2$ and then measuring $\text{H}_2^{34}\text{SO}_4$. ^{34}S is used to discriminate between H_2SO_4 produced from OH and ambient H_2SO_4 – which is also measured.

HO₂ and HO₂+RO₂

HO₂ and HO₂+RO₂ are also measured by converting them to H_2SO_4 via reaction with SO_2 and NO then measuring H_2SO_4 .

HO₂ is measured in a low O₂, relatively high NO environment, in which most RO₂ are not converted to H_2SO_4 .

HO₂ + RO₂ is measured in a high O₂, relatively low NO environment, in which both are converted to H_2SO_4 with high efficiency.

H₂SO₄

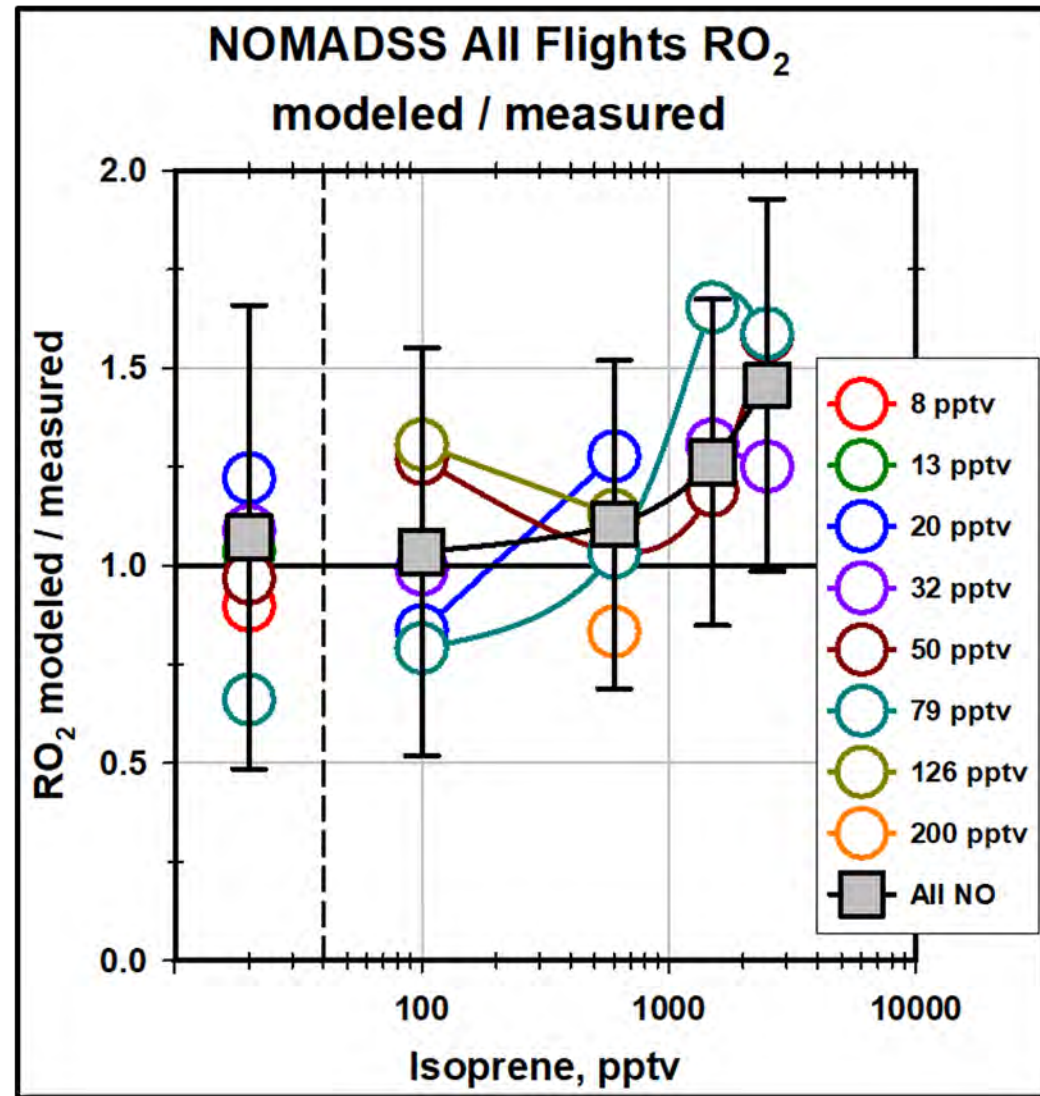
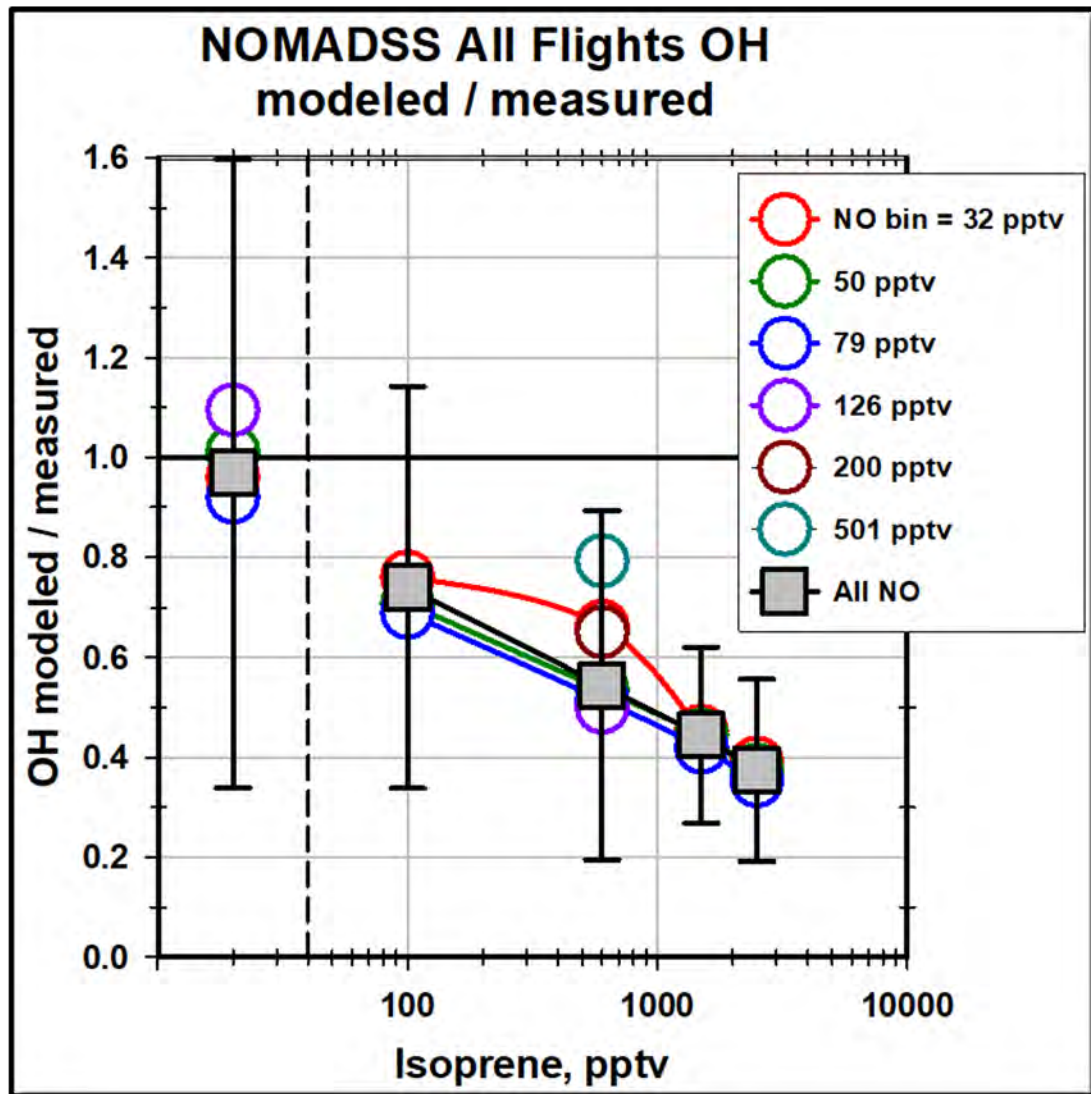
H₂SO₄ is measured using nitrate Chemical Ionization Mass Spectroscopy (CIMS)

Detection Limits

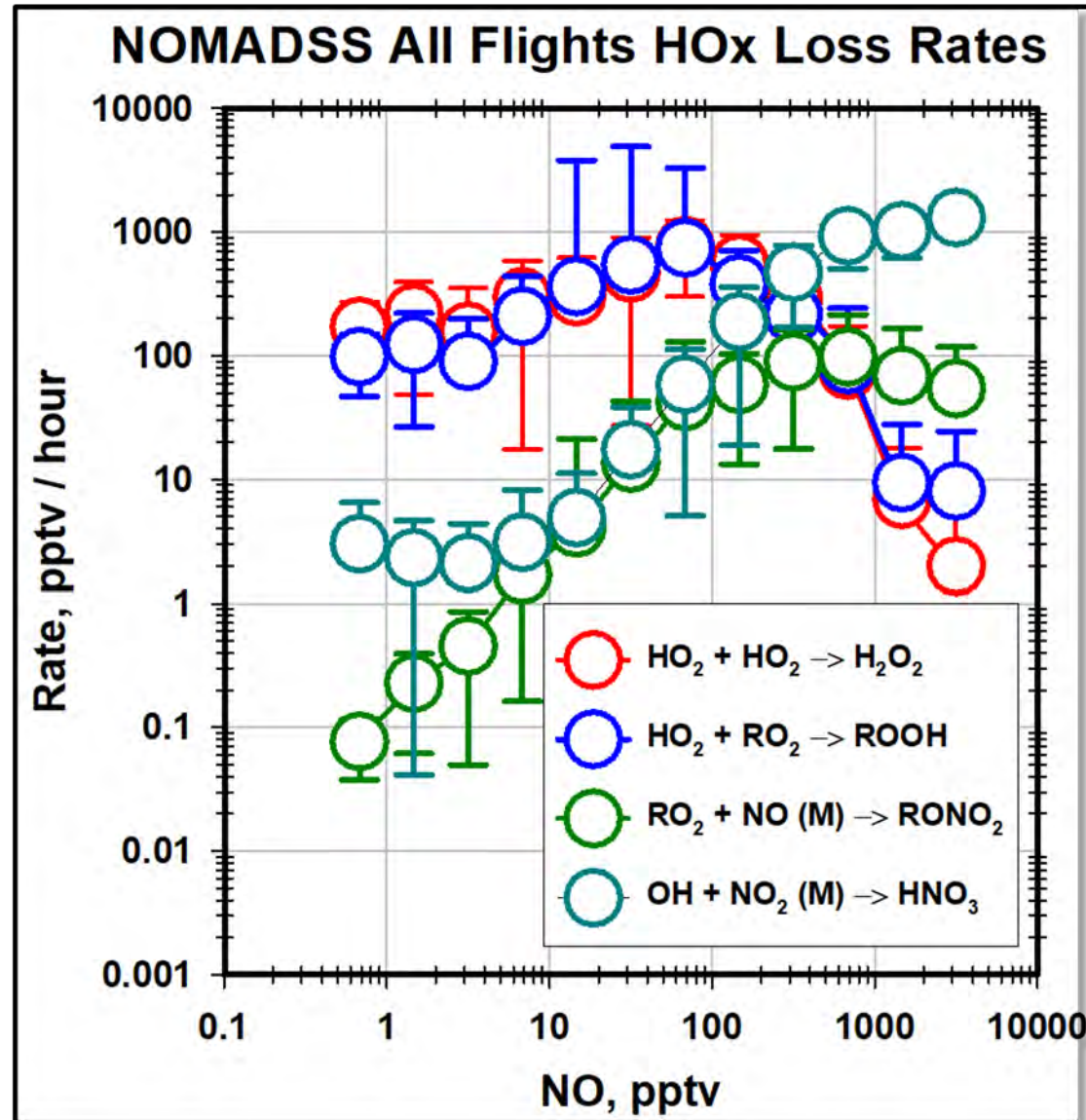
OH, H₂SO₄ - 3×10^5 molecule cm⁻³ for 30 sec measurement

HO₂, HO₂+RO₂ - 5×10^5 molecule cm⁻³ for 60 sec measurement

Box model underpredicts OH and overpredicts RO₂ at high isoprene



Direct Calculation of Radical Termination Rates



In Situ Measurements of NO, NO₂, and O₃

- For NO, reagent O₃ is added to the flow of ambient air.
- Photons are counted using dry-ice cooled PMT.
- For NO₂, first convert to NO in cooled photolytic cell.
- For NO_y, first convert to NO on heated gold catalyst.
- For O₃, reagent NO is added to flow of ambient air.

People you might see around the NO_x / O₃ /carbon rack:



Ale Franchin (PI)



Courtney Owen



Kirk Ullmann



Kirk Lesko

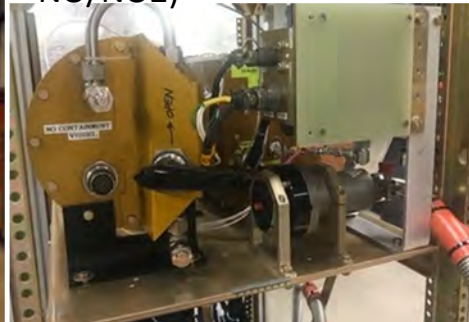


Teresa Campos

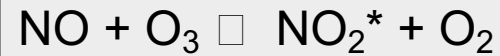
NCAR 2-Channel instrument
(NO & NO₂, integrated with O₃)



HAIS O3
(integrated with
NO/NO2)



3 sample flows, each
employing the
chemiluminescent
reaction:



Nominal detection
limits:

NO: 10 pptv
NO₂: 20 pptv
O₃: 50 pptv

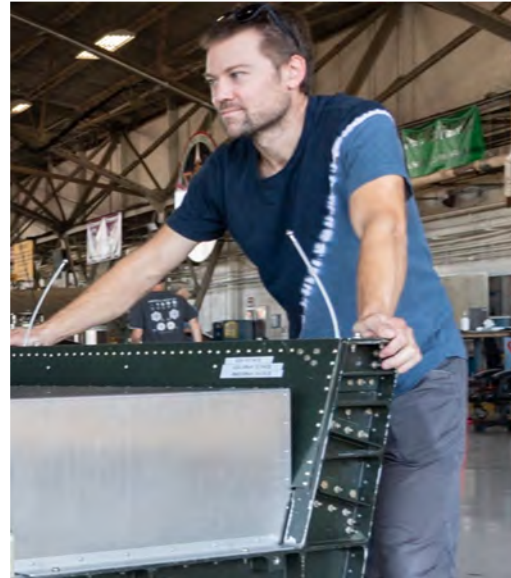
Data reported at 1s
time resolution

Shared double rack with ACOM Carbon instrumentation,
total installation taking up one station in C-130

NO_x/NO_y and O₃ measurements on the DC-8 for AEROMMA



Kristen Zuraski



Drew Rollins



Eleanor Waxman



Jeff Peischl

NO-LIF Technique / O₃ Chemiluminescence Technique

NO/NO₂/NO_y Technique

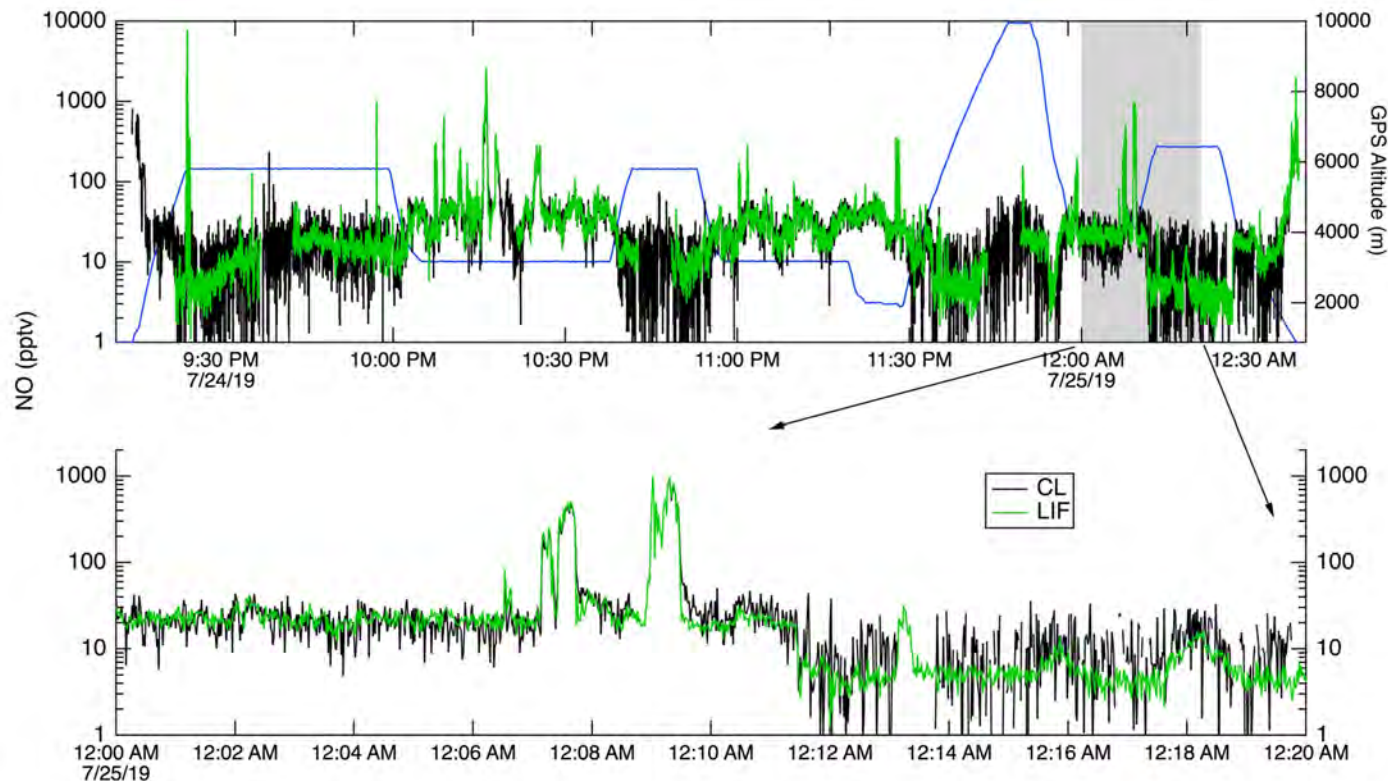
- Laser excitation @ 215 nm
- Fluorescence collection ~250-260 nm
- 1 s detection limit: < 1 ppt
- 3 channel instrument for continuous measurements of NO, NO₂, NO_y

NO - LIF Heritage

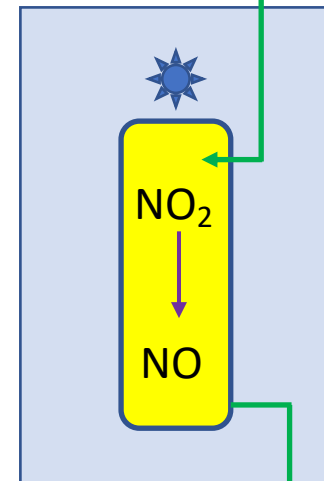
- FIREX-AQ 2019 (DC-8)
- SABRE 2022 (WB-57)
- ACCLIP 2022 (WB-57)

O₃ CL Instrument:

- O₃ + NO -> light
- High precision
- No interferences



NO₂ converter



395 nm LED
Photolytic
converter held in
cold winglet

LIF NO
Instrument

DC8 Cabin

NO_y / O₃ Science Foci

Marine

- Fluxes of NO and O₃
- NO_y closure in the MBL
- NO_x gradients in coastal regions and impact of peroxy radical fate on oxidation product distributions

Continental

- NO_x sources in urban areas
- Impact of NO_x on O₃ production
- Fate of peroxy radicals and impact of NO_x
- UT NO₂ / NO ratios and lightning NO_x



**Colorado
State**
University

**Department of
Atmospheric Science**

Contact:
ilana.pollack@colostate.edu

Gas-phase Ammonia Measurements for AEROMMA on the DC-8



Ilana Pollack
*PI, Instrument
Team Lead*



Emily Lill
*PhD Candidate,
Team Member*



Emily Fischer
*co-PI,
Team Member*



Jeff Pierce
*co-PI,
Team Member*



Gas-phase NH_3 on the DC-8 using a QC-TILDAS

Instrument features:

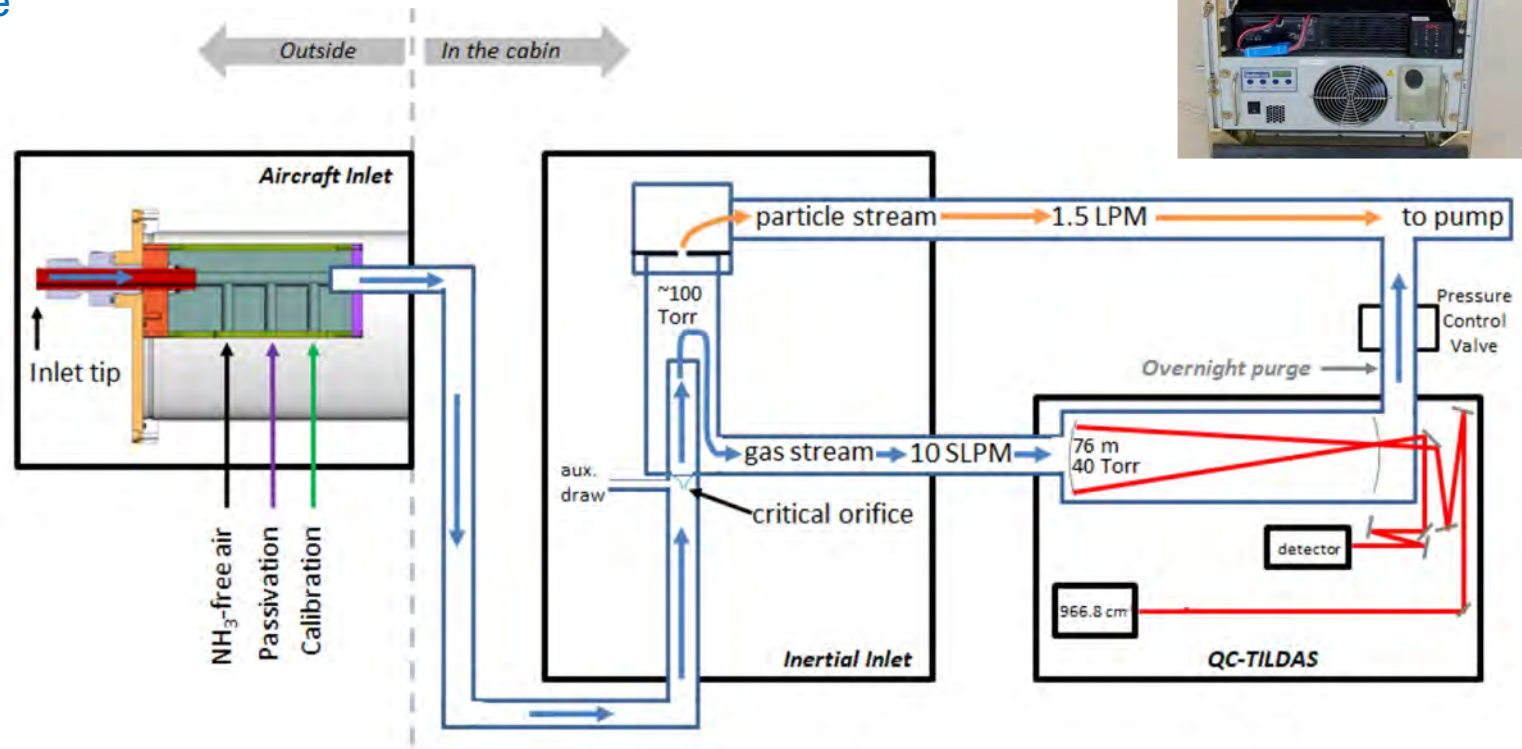
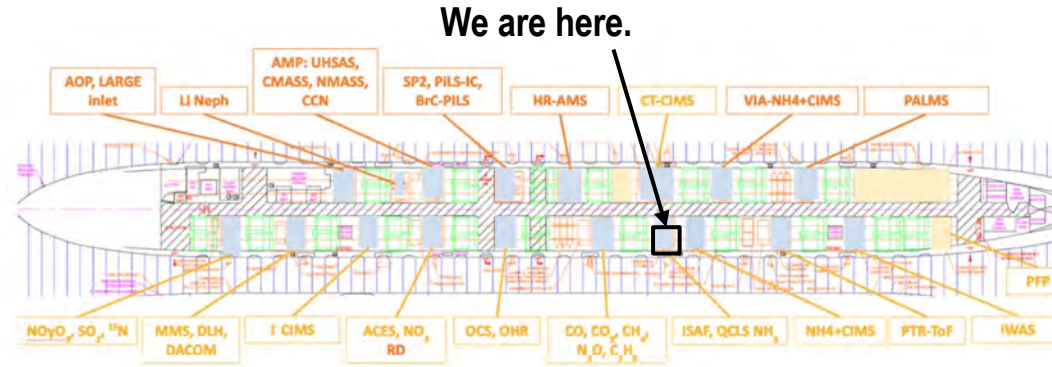
- Commercial Aerodyne QC-TILDAS mini-CS
- Filterless, aerodynamic separation of particles
- Vibration applied to laser obj. for reduced noise
- Isolation mounted for reduced motion sensitivity
- Active continuous passivation and heating of sampling surfaces for improved time response

Instrument performance*:

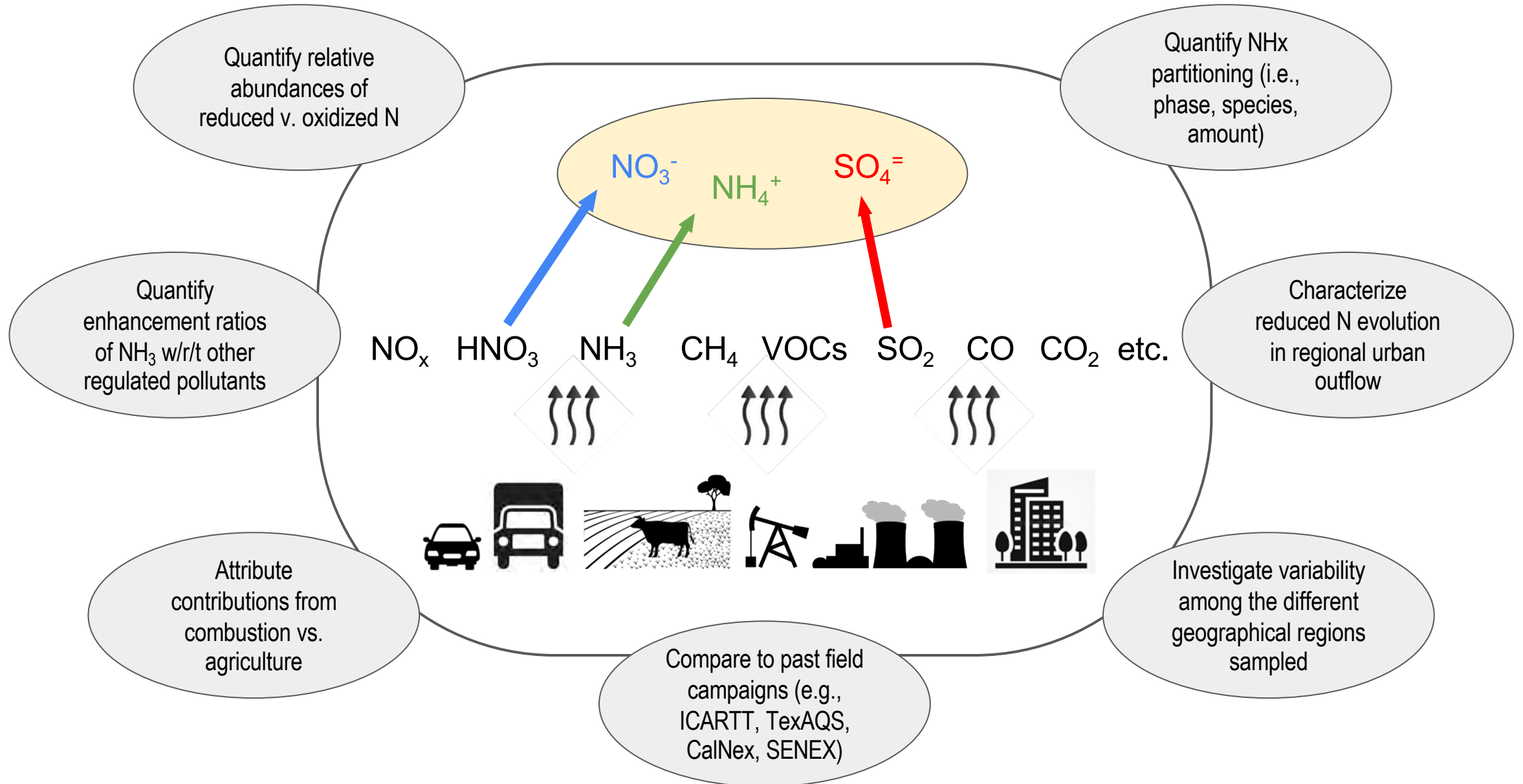
Meas. Freq.: 10 Hz sampled, 1 Hz reported
 Time Res. (t_{90}): <1 s with, ~2 s w/out passivant
 Precision: <60 ppt @ 1 Hz
 Detection Limit: <200 ppt @ 1 Hz
 Accuracy: 12%
 *As determined in flight on the NSF/NCAR C-130.

Key reference:

Pollack et al. (2019), AMT., 12, 3717–3742,
 DOI: [10.5194/amt-12-3717-2019](https://doi.org/10.5194/amt-12-3717-2019).



Ammonia for AEROMMA Science Goals





OH REACTIVITY INSTRUMENT

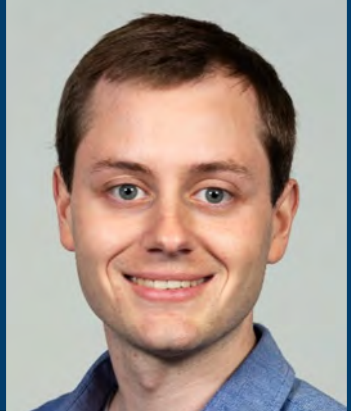
Forschungszentrum Jülich
Institute of Energy and Climate Research



Hendrik Fuchs
Instrument PI



Anna Novelli
Scientist



Aaron Stainsby
PhD Student

SEPTEMBER 27, 2022

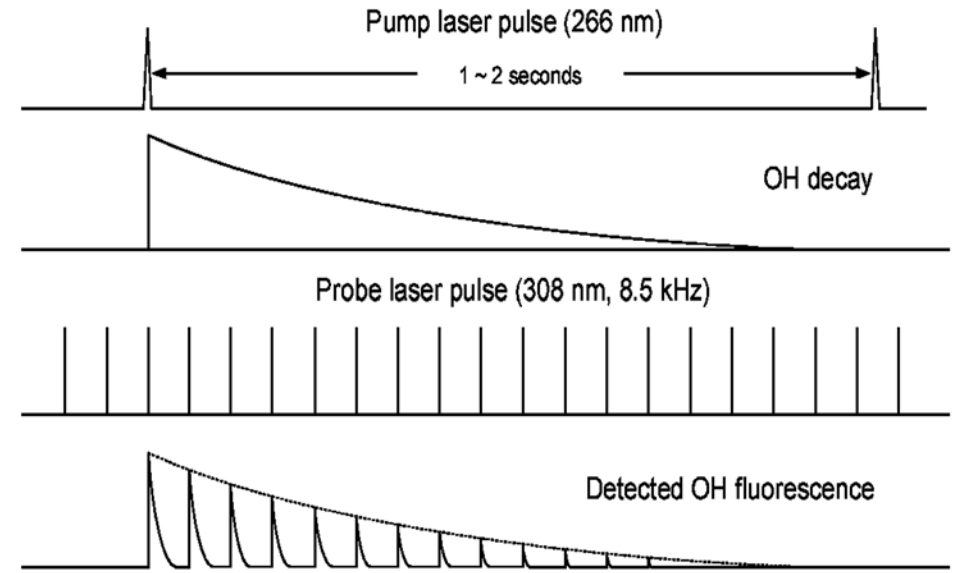
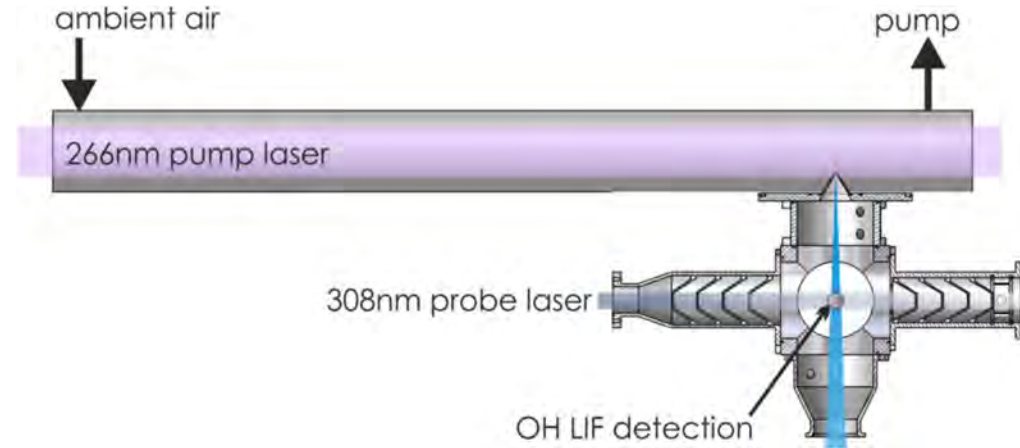
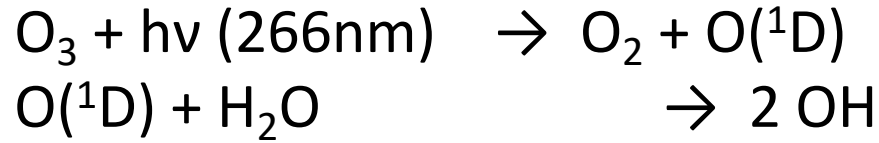
Mitglied der Helmholtz-Gemeinschaft



Contact: H.Fuchs@fz-juelich.de



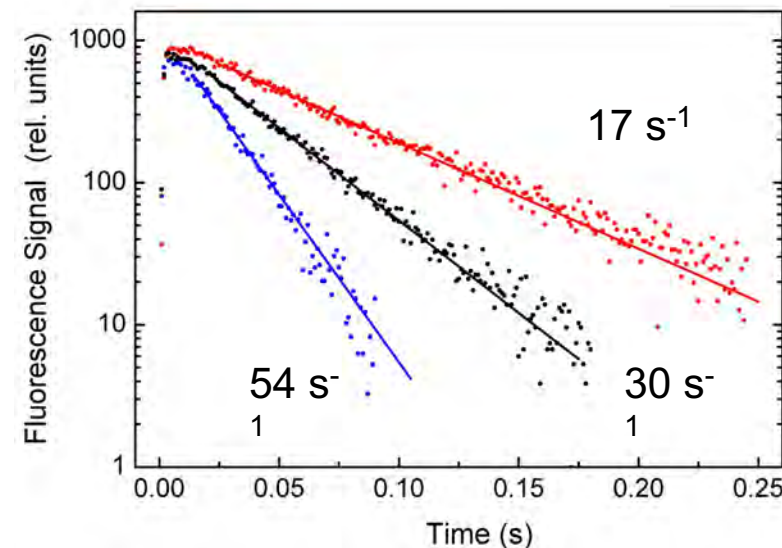
LIF OH-REACTIVITY



$$[\text{OH}](t) = [\text{OH}]_0 e^{-(k_{\text{OH}}t)}$$

□ time

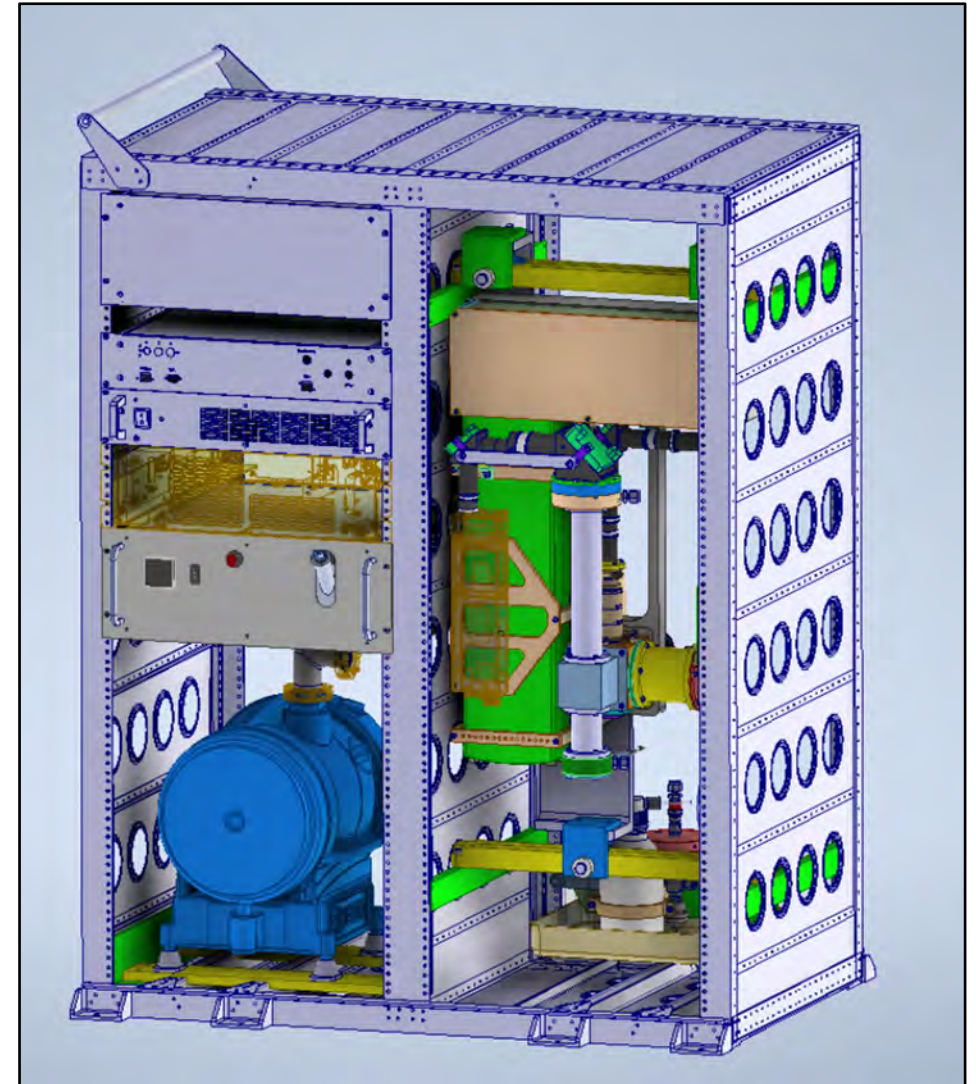
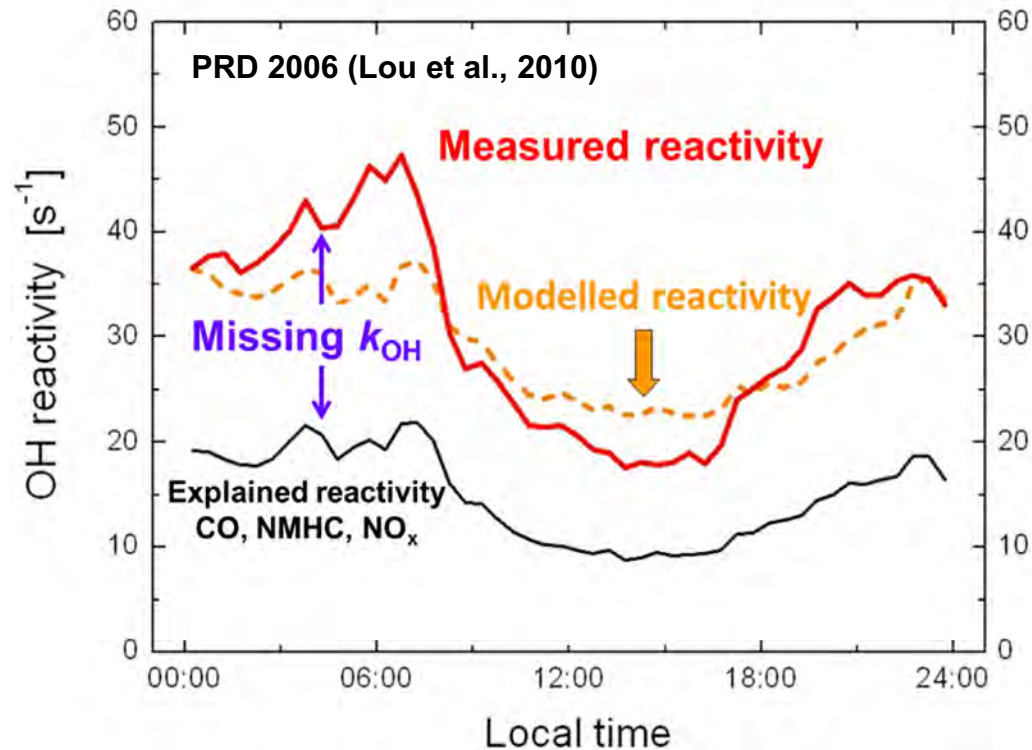
	k_{OH} instrument
Precision	0.2 s^{-1}
Accuracy	$(5 - 10) \% \pm 0.7 \text{ s}^{-1}$ for $\text{NO} < 20 \text{ ppb}$
Limit of detection	$2\sigma = 0.6 \text{ s}^{-1}$



SCIENCE GOALS AND FOCI

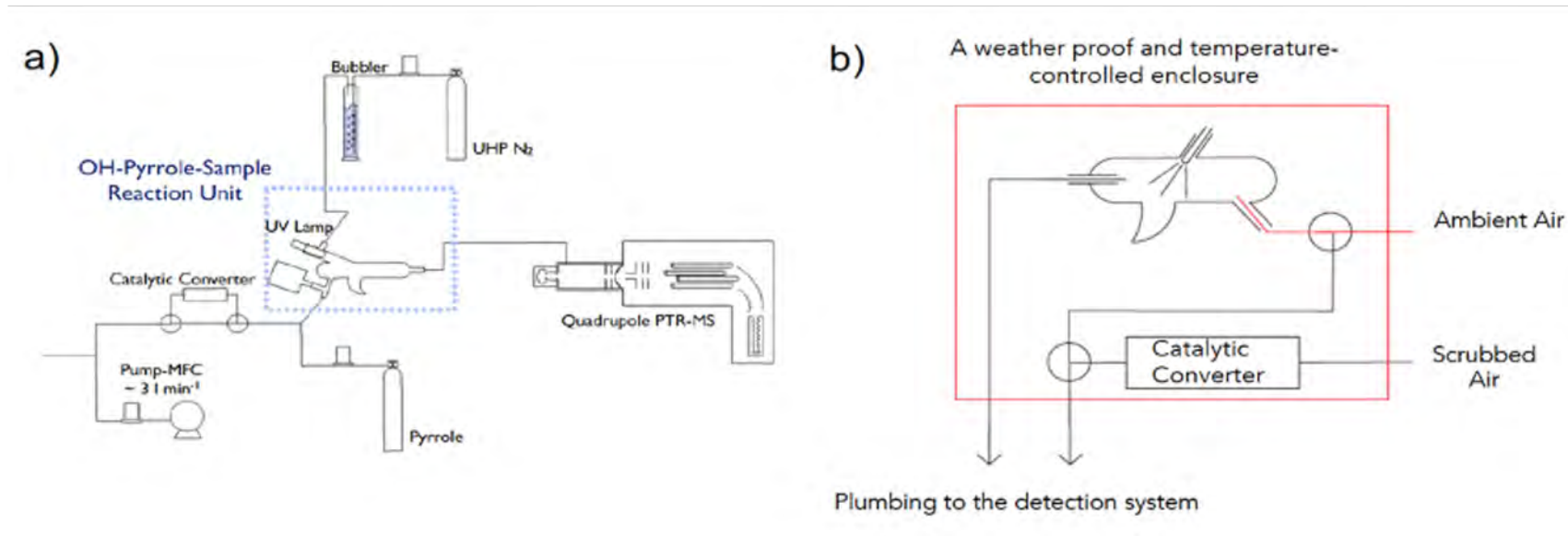
Test completeness of $k_{OH}^{\text{calculated}}$ by comparison with k_{OH}^{measured}

$$k_{OH} = \underbrace{\sum k_{OH+X_i} [X_i]}_{\text{known species } X_i} + k_{OH}^{\text{missing}}$$



Determination of Instantaneous Ozone Production Regime with total OH reactivity measurements in the New York Metropolitan Area during the AEROMMA campaign

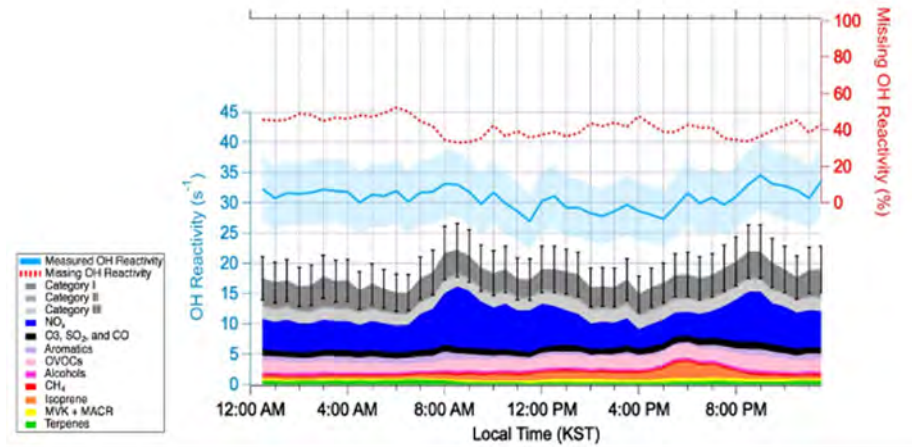
UC Irvine - Saewung Kim, Alex Guenther, Sanjeevi Nagalingam, Gracie Wong, Julie Devlin, and Katherine Parodre



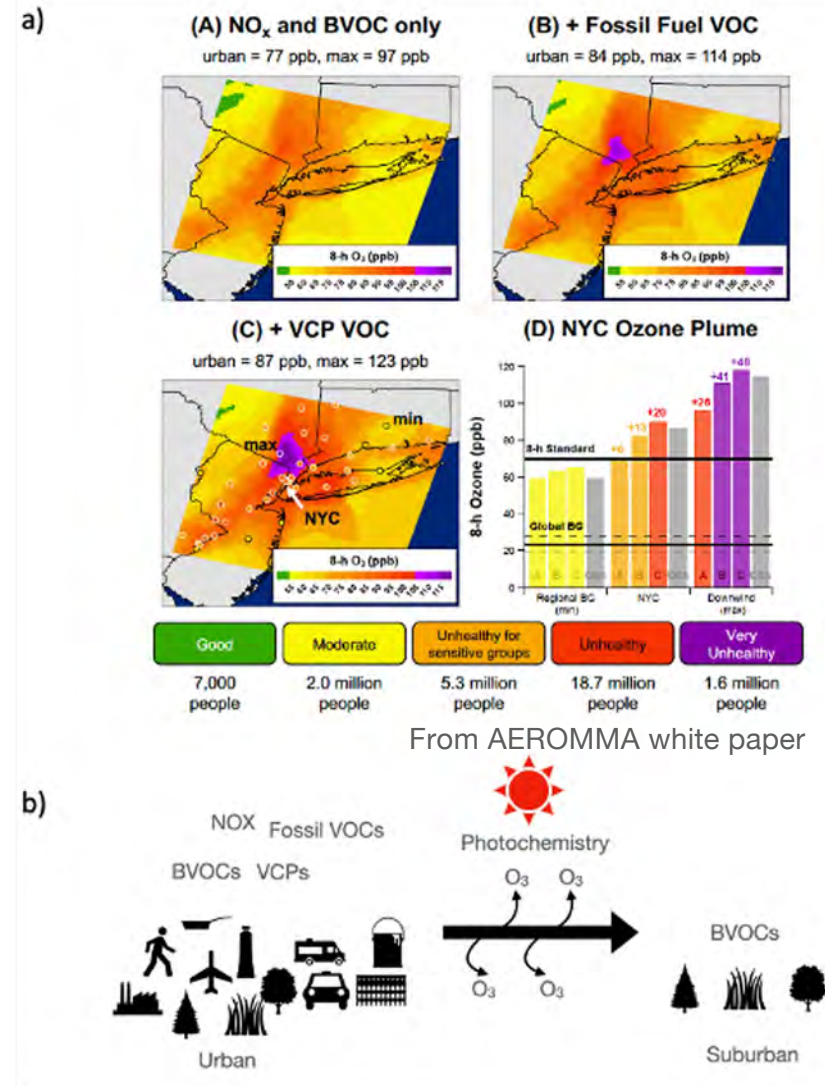
- Ground OH reactivity Observations using the Comparative Reactivity Method (Sinha et al., 2009) with an updated inlet system
- Possible ground observations for the CA deployment

Specific science questions

Sanchez et al., (2020)



- Identifying any unmeasured/unidentified compounds
- Source appointments for the unaccounted/unknown
- Wholistic identifications for the ozone production regimes



Additional Observations

TOF CIMS Measurements (SF_6) to quantify radical precursors: HONO, HO_2NO_2 , and HCl
BVOC
distribution

Drexel ECHAMP (HO₂ & RO₂) at CUNY ASRC (NYC-METS)



*Ezra Wood
Instrument PI*



*Tauhidur Rahman
Grad Student*



*Khaled Joy
Grad Student*



*Contact:
Ezra.Wood@drexel.edu*

Drexel Ethane Chemical Amplifier HO₂ + RO₂ at CUNY Advanced Science Research Center (NYC-METS)



Targeted Species

“Total” Peroxy Radicals.

Only detects peroxy radicals that react with NO to form NO₂ and propagate HOx cycle

Instrument Performance

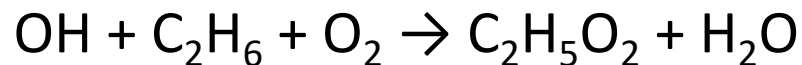
Time resolution: 2 minutes

Detection limit: 2-4 ppt

Precision: 1-2 ppt

Accuracy: 20 - 30%

Measure
NO₂



← Air

NO, C₂H₆

References:

Anderson et al. (2019) *ACP* 19.5: 2845-2860

Wood et al. (2017) *ES&T Letters* 4.1: 15-19

Anderson et al. (2021), *ES&T* 55.8: 4532-4541



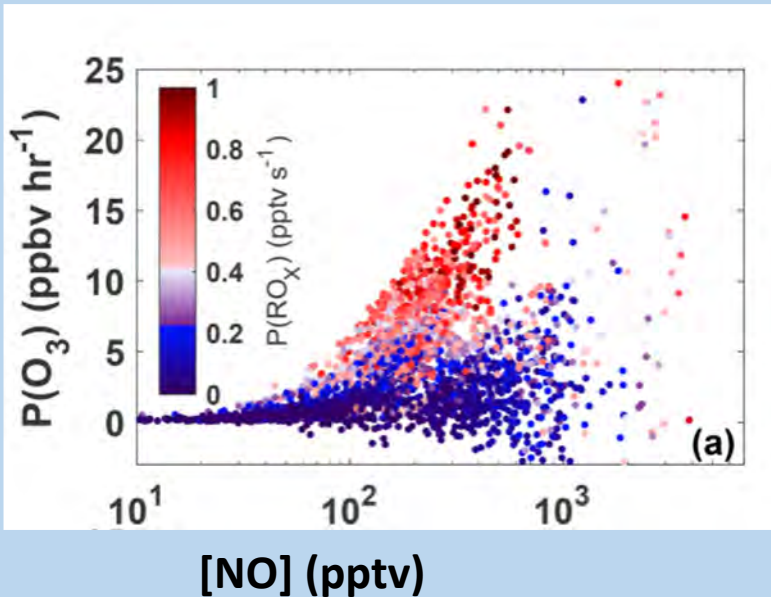
Science Goals & Questions



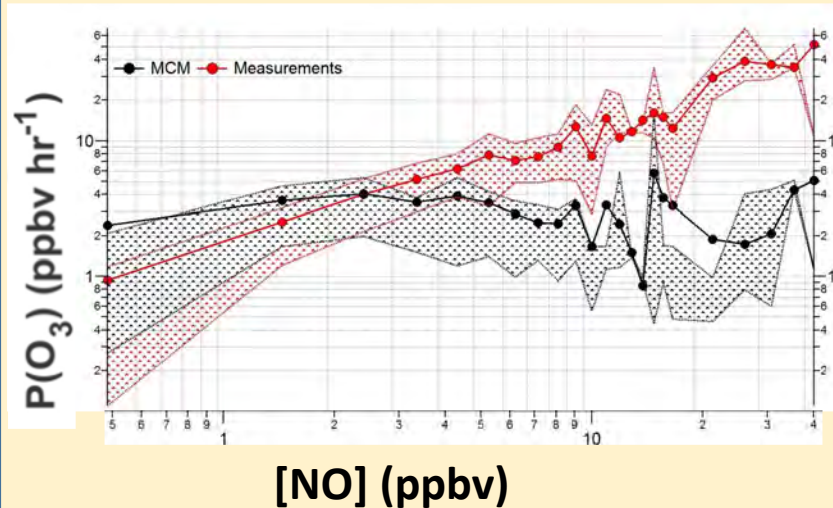
Characterize Instantaneous Ozone Production Rate $P(O_3)$ and dependence on NO_x & VOCs (especially VCPs)

Determine Instantaneous Ozone Production Rates:
 $P(O_3) = k([HO_2] + [RO_2])[NO]$

Example from San Antonio, 2017:



Do we understand high- NO_x ozone chemistry? Investigate with 0-D models (collaborative)



Large model-measurement discrepancies observed in London (Whalley et al. ACP 2018), Beijing, Wangdu, Colorado.

Bonus activities:

- Provide NO and NO_2 measurements (chemiluminescence/CAPS)
- Quantify on-road NO_x emission factors (C-balance method)
- Provide photolytic HONO calibration source for I-CIMS (Lindsay and Wood, AMT 2022)

York University - Reactive Nitrogen Measurements



**SEYED DANIAL NODEH
FARAHANI**

ToF-CIMS Instrument (acetate)

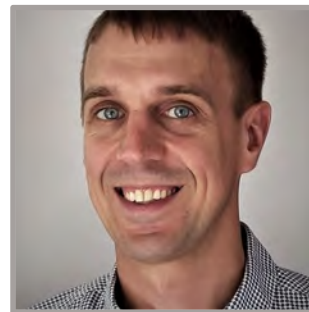
PhD Student
Department of Chemistry
York University



MOXY SHAH

Nitrogen Oxides
Instrumentation

MSc Student
Department of Chemistry
York University



TREVOR VANDENBOER

Team Leader

Assistant Professor
Department of Chemistry
York University

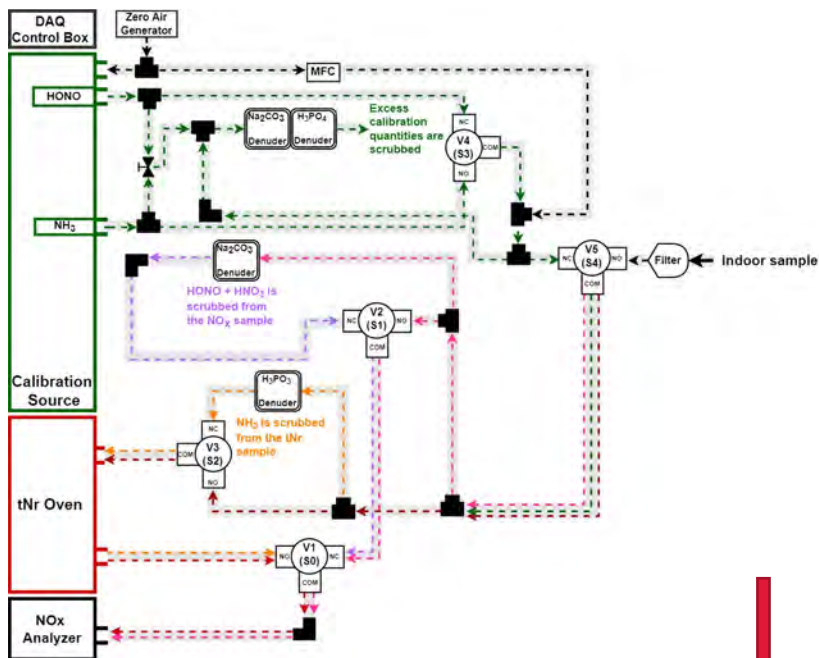


**PROSPECTIVE GROUP
MEMBER**

ToF-CIMS and Total Reactive
Nitrogen Instrumentation

Graduate Student and/or Postdoc
Department of Chemistry
York University

York University –H-ToF-CIMS (Acetate) and Total-N_r



- DAQ Control System (LabJack & LabView)
- Laptop Area
- HONO/NH₃ Calibration Source (Lao et al. 2020)
- Total Reactive Nitrogen Oven (Stockwell et al. 2018)
- NO_x Analyzer (American Ecotech, Serinus 40)
- Pumps (2)
- T-Slotted Framing Rails & Casters (McMaster Carr)

Targeted Species

Total N_r

Gases: HONO+HNO₃, NR₃, NO, NO₂

Particulate: All condensed N_r

Time resolution: 5 – 20 minutes

Detection limits: 0.4 ppb for 1 min

Precision: signal/analyte dependent

Accuracy: ~±20 %

H-ToF-CIMS

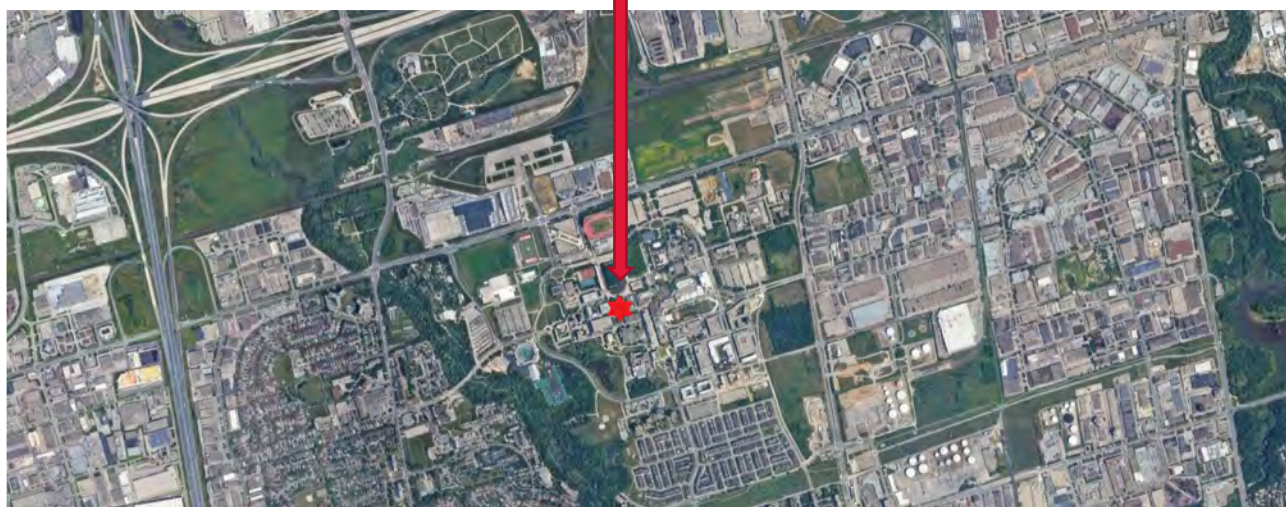
Inorganic Acids: HCl, HONO, HNO₃

Organic Acids

Persistent Acids (PFCAs): TFA

Haloacetic acids: monochloroacetic acid

Performance: Currently under determination



York University –H-ToF-CIMS (Acetate) and Total-N_r



Science Goals

Total N_r

- NO_x contributions to O₃ and radical reservoir formation
- Urban gas and particle phase N_r budgets
- Inorganic vs organic N in PM
- Validate through CIMS and AIM-IC measurements

H-ToF-CIMS

- Impacts of HONO on radical budget
- Acid impacts on secondary inorganic PM formation
- Formation and fate of persistent halogenated acids in urban air



Ozondesondes during STAQS



Paul Walter
St. Edward's University

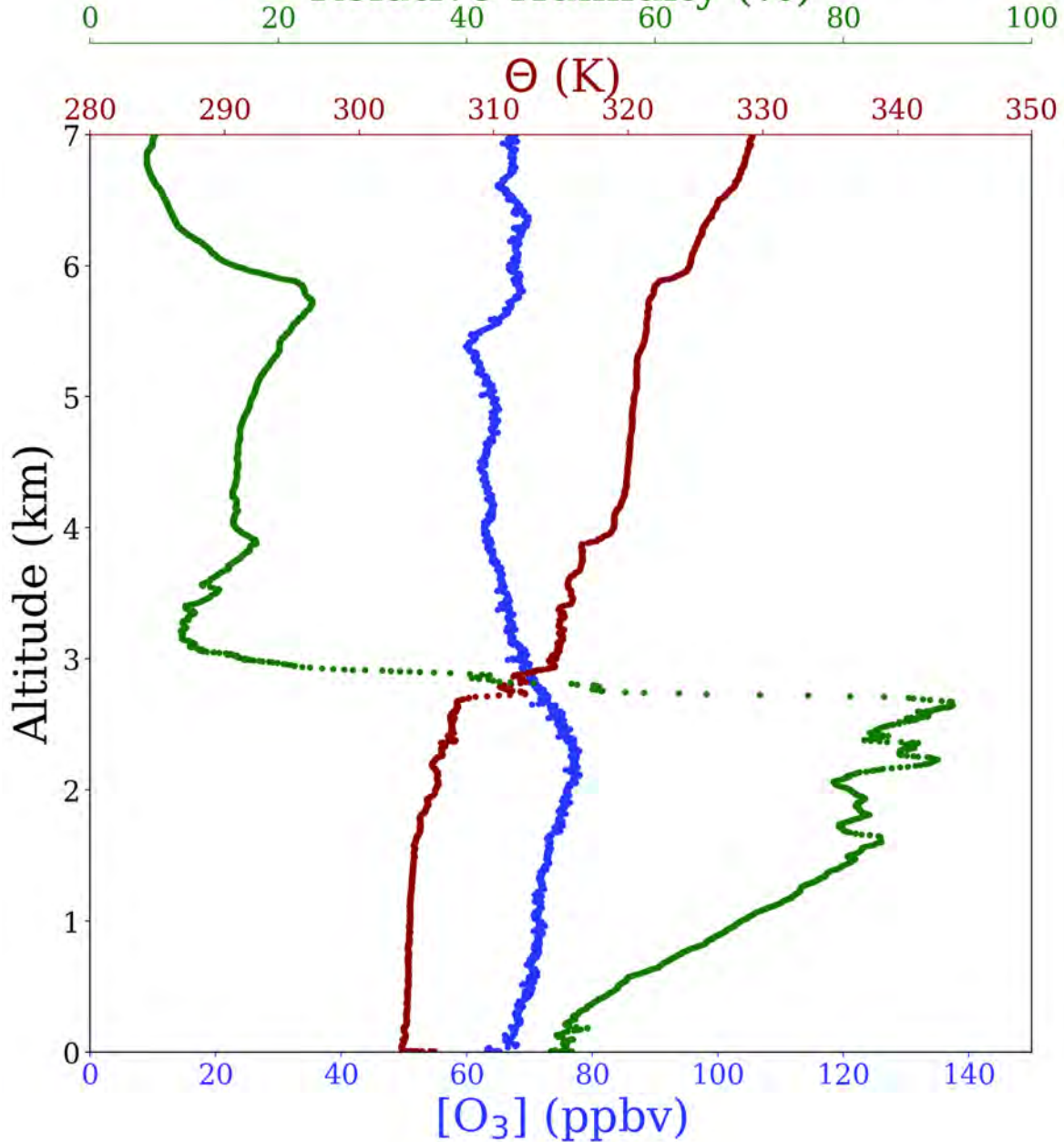


John Sullivan
NASA Goddard



09 Sept 2021 Galveston Bay (20:32 UTC)

Relative Humidity (%)



Ozonesondes during STAQS

Balloon-based Instruments:

En-Sci 2Z Ozonesonde

iMet-4RSB Radiosonde

Data Collected:

Vertical Profiles of O₃ and meteorological data

Altitude range:

Surface to 24-30 km depending on balloon size

Rise rate of ~5 m/s

Instrument Performance:

Ozone

Accuracy: 5-10%

Response time: 25 s

(vertical resolution ~125 m)

Time Resolution: 1 Hz

Ozonesondes during STAQS

50-60 ozonesondes near Long Island Sound

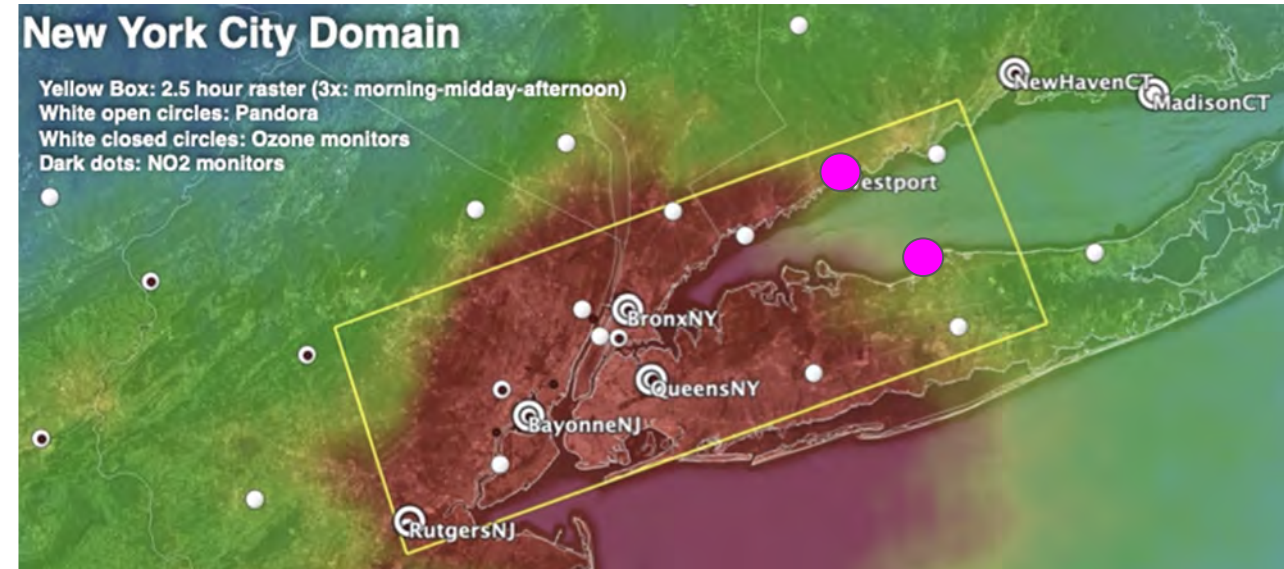
- ~20 collocated at each TOLNet ground-based ozone lidar site
 - NASA GSFC ozone lidar (Flax Pond, NY)
 - NASA LaRC ozone lidar (Westport, CT)
- 10-20 from other locations

Validation of

- ground-based & airborne ozone lidars
- TEMPO
- Pandoras

Characterize ozone episodes

- Boundary layer and residual layer ozone
- Identification of possible ozone transport
- Boundary Layer heights
- Vertical profiles of U and V winds



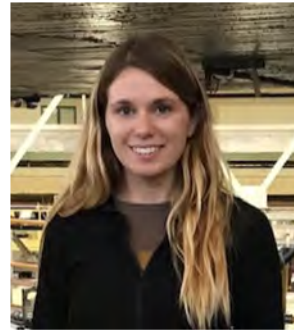
Ground-based ozone lidar locations shown in pink. Image courtesy of Laura Judd.



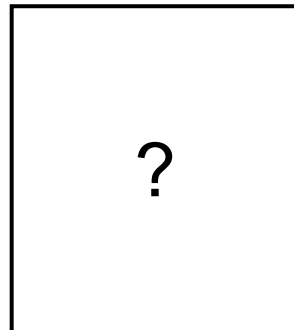
NOAA Nitrogen Oxide Cavity Ring Down Spectrometer (NO_xCaRD)



4 Channel, one color CRDS instrument for NO, NO₂, NO_y, O₃



Kristen Zuraski



1. Direct detection of **NO₂** @ 405

1. Convert NO to NO₂ in excess O₃ to measure NO_x



2. Convert O₃ to NO₂ in excess NO to measure O_x



3. Thermally dissociate NO_y to NO₂



NO₂: ≥ 20 pptv, 5% accuracy / 1 Hz

NO, O₃: ≥ 50 pptv, 7% accuracy / 1 Hz

NO_y ≥ 20 pptv, 12% accuracy / 1 Hz

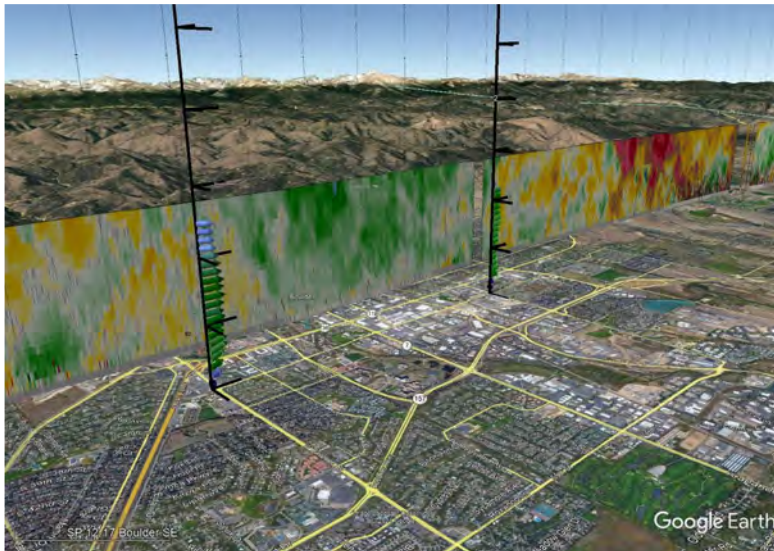
Science Objectives: Coastal Urban Plume Dynamics Study (CUPiDS)



- Vertical profiles and cross boundary layer gradients of NO_2
- Air mass aging and photochemistry (NO_x/NO_y)
- Urban ozone production efficiency

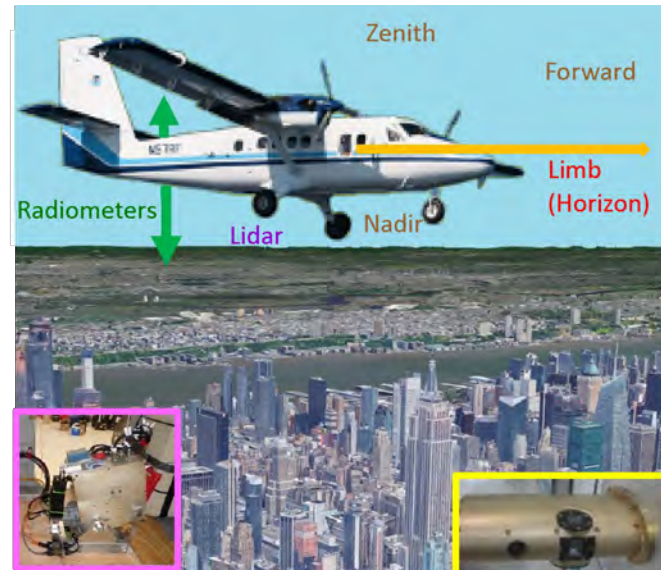
Scanning Doppler Lidar (CSL)

- Wind, Turbulence and Aerosol profiles
- Boundary Layer Depth



MAX-DOAS (CU – AC4 supported)

- NO_2 , Formaldehyde, Glyoxal, IO profiles
- Surface albedo



In-Situ Airborne (CSL, GML)

- GHG, NO_2 , NO, NO_y , O_3
- Temp, Press, RH

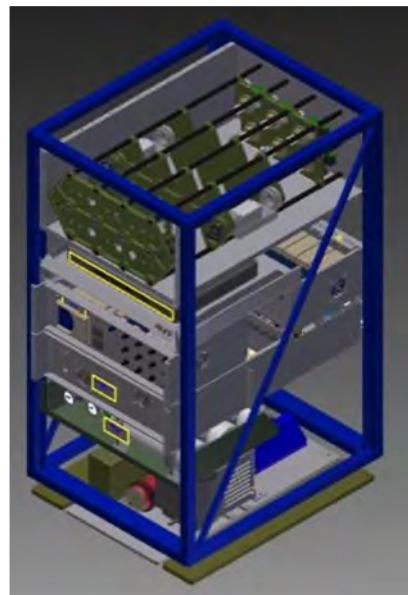
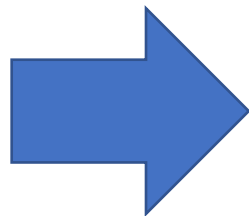




NOAA Nighttime Nitrogen Oxides Cavity Ring Down Spectrometer (NNOx)



P-3 Instrument: Developed 2004



Single Bay: 2022



Wyndom
Chace



Carrie
Womack

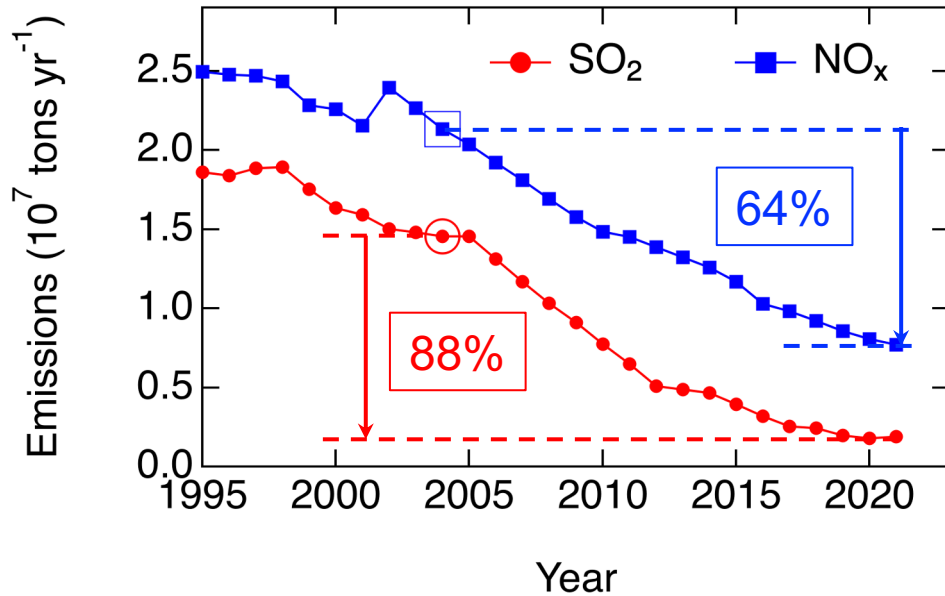
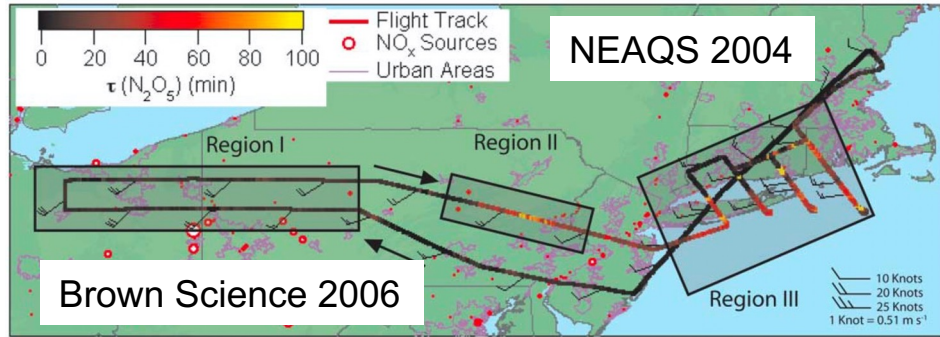
4 Channel, 2 Color CRDS
Instrument for NO_3 , N_2O_5 , NO_2 , O_3

- 2 Channels @ 662 nm for detection of NO_3 (direct) and N_2O_5 (thermal dissociation to NO_3)
 $\text{N}_2\text{O}_5 + \Delta (140^\circ\text{C}) \rightarrow \text{NO}_2 + \text{NO}_3$
 NO_3 by 662 nm optical extinction, sensitivity ≥ 0.2 pptv / 1 Hz
- 2 Channels @ 405 nm for detection of NO_2 (direct) and O_3 (via conversion to NO_2)
 $\text{O}_3 + \text{NO} (\text{excess, } \sim 20 \text{ ppm}) \rightarrow \text{NO}_2 + \text{O}_2$ (or sum of $\text{NO}_2 + \text{O}_3 = \text{O}_x$)
 NO_2 by 405 nm optical extinction, sensitivity ≥ 20 pptv / 1 Hz

Science Objectives

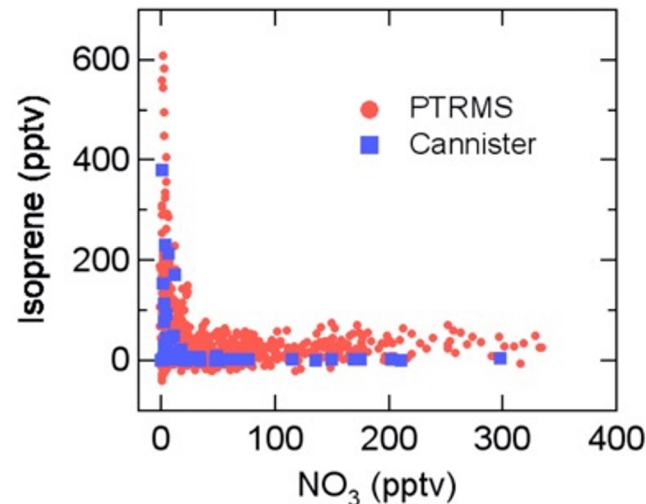
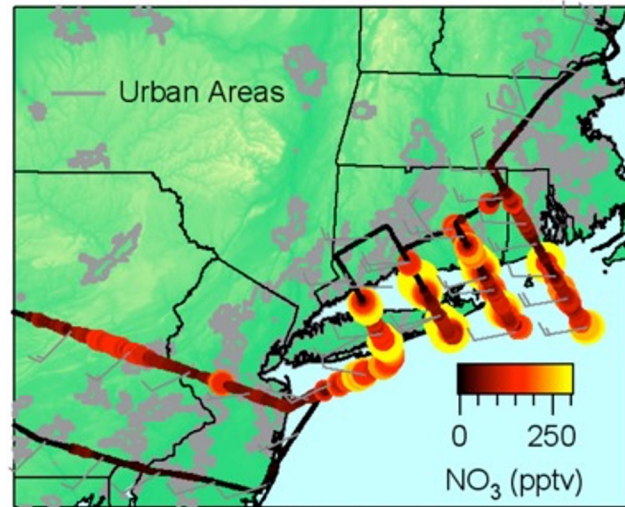
Limited time for flying after dark, and no dedicated night flights !

1. N_2O_5 heterogeneous uptake

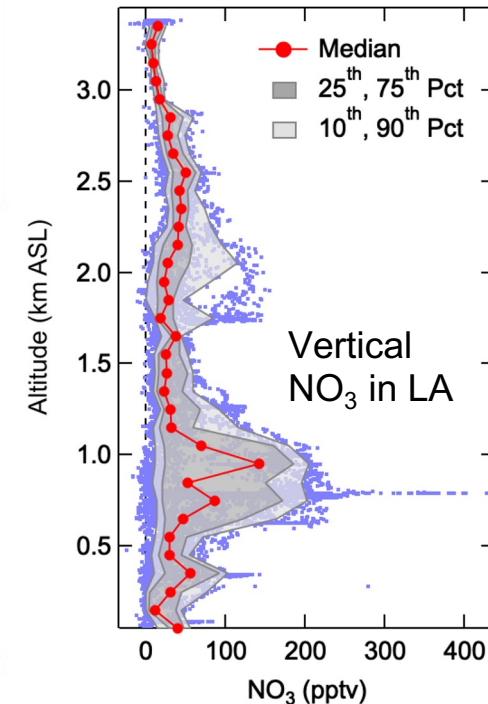
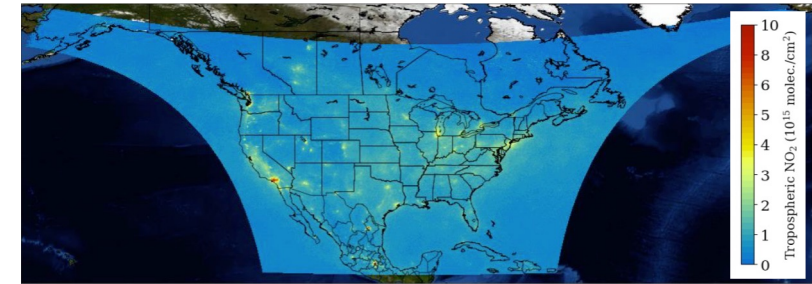


Change in sulfate N_2O_5 uptake ?

2. Isoprene Oxidation



3. Satellite NO₃ ?



Red channels on TEMPO and GCAS (G-V)

Twilight measurement of nighttime species

Reactive Nitrogen, Ozone, and Radicals

Glenn M. Wolfe (Moderator)
NASA Goddard Space Flight Center

AGES Meeting
28 September 2022

Speaker	Measurement	Mission
Glenn (for Lee Mauldin)	RO _x , H ₂ SO ₄	GOTHAAM
Glenn (for Ale Franchin)	NO _x , O ₃	GOTHAAM
Andrew Rollins	NO _x , NO _y , O ₃	AEROMMA
Ilana Pollack	NH ₃	AEROMMA
Aaron Stainsby	OH Reactivity	AEROMMA
Saewung Kim	OH Reactivity	NYC-METS
Ezra Wood	HO ₂ + RO ₂	NYC-METS
Trevor VandenBoer	Acids, Total N _r	THE CIX
Paul Walter	Sondes	STAQS
Steve Brown	NO _y , O ₃	CUPiDS
Steve Brown	“Night NO _x ”	AEROMMA